

# The 479th Geodynamics Seminar

## The effect of sulfur on carbon solubility and partitioning in the alloy-silicate systems: Implications for core-mantle fractionation of carbon and sulfur during accretion of Earth

**Dr. Kyusei Tsuno**

Research Scientist/Lab Manager, Rice University

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### Abstract

Constraining the carbon (C) fractionation between the silicate magma ocean and core-forming alloy liquid during early differentiation of Earth and terrestrial planets is required to determine the origin and present-day distribution of C between planetary reservoirs such as exosphere, mantle and core [1]. Experiments on the metal-silicate partitioning of C have shown that preferential fractionation of C into the alloy liquid would have left the bulk silicate Earth (BSE) devoid of C [2-4]. It has been recently proposed that merger of a sulfur (S)-rich differentiated body into the proto-Earth could have supplied almost the entire C budget of the present-day BSE [5]. However, systematic experimental data at high pressures on the effect of S on C solubility in alloy liquid and the effect of S on partitioning of C between Fe-rich alloy liquid and silicate melt are not fully known. We have performed multi-anvil experiments at 6–13 GPa and 1800–2000 °C to examine the effects of S and Ni on the solubility limit of carbon in Fe-rich alloy liquid and simultaneous determination of C and S partitioning between alloy liquid and silicate melt. The composition of alloy liquid, including carbon, was determined using EPMA. Major elements and volatiles (C and H<sub>2</sub>O) in the silicate glass were obtained using EPMA and SIMS, respectively. The obtained data were used to quantify the effect of S on the distribution of C between the silicate magma ocean and core-forming alloy liquid of Mars-sized impacting body. A simple two-stage equilibrium core formation model was tested to determine whether the merger of an S-rich impacting body to a volatile-depleted proto-Earth can satisfy C and S contents as well as C/S ratio of the BSE [6].

[1] Dasgupta (2013) *RiMG*. [2] Dasgupta et al. (2013) *GCA*. [3] Chi et al. (2014) *GCA*. [4] Li et al. (2015) *EPSL* [5] Li et al. (2016) *Nature Geo* [6] Hirschmann (2016) *Am Mineral*.

Contact : Dr. Nishi (e-mail: [nishi@sci.ehime-u.ac.jp](mailto:nishi@sci.ehime-u.ac.jp))